

PREPARATION OF AMINO DERIVATIVES OF SOME NITROGEN-CONTAINING HETEROCYCLES

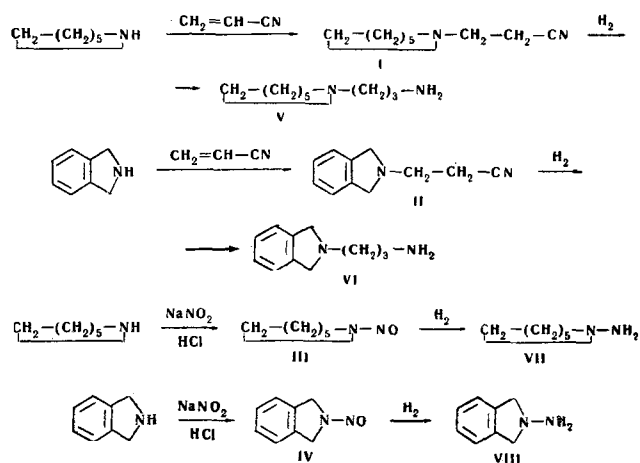
T. A. Sladkova and E. A. Chernyshev

*Khimiya Geterotsiklicheskikh Soedinanii*, Vol. 4, No. 1, pp. 140-141, 1968

UDC 547.758:542.941.4.7'942.4

This paper presents a description of the preparation of compounds containing a primary amino group in the side chain and a tertiary nitrogen atom in the ring, starting from hexamethyleneimine and isoindoline.

At present a large number of physiologically active compounds with nitrogen-containing heterocycles in their molecular structure are known. In particular, pharmacological preparations are obtained from hexamethyleneimine as well as from isoindoline [1, 2]. N-Alkyl derivatives of hexamethyleneimine [3, 4] and of acylindolines [5] possess repelling activity. Some esters of hexamethylenedithiocarbamic acid are poisonous to soil nematodes [6]. Hexamethyleneimine, obtained as a secondary product in the manufacture of hexamethylenediamine, is an available initial compound. The simple methods which have been proposed for the preparation of primary amines, with nitrogen-containing five- and seven-membered heterocycles, may prove interesting in the search for new physiologically active preparations. The reduction of compounds I-IV was formerly accomplished either with  $\text{LiAlH}_4$  [7] or under pressure with hydrogen and reduced nickel or cobalt catalyst in ethanol in the presence of ammonia. More satisfactory results were obtained by catalytic hydrogenation.



EXPERIMENTAL

**N-(β-Cyanoethyl)hexamethyleneimine (I).** The reaction was conducted by the same method developed for obtaining N-(β-cyanoethyl) piperidine [8]. To hexamethyleneimine was added a small excess (11-11.5%) of acrylonitrile, and after the evolution of heat had ceased, the mixture was heated in a sealed glass ampoule for 6-16 hr at 100° C. Yield of I, 85-90%, colorless, transparent liquid boiling at 94°-95° C (4.5 mm), at 113° C (8.5 mm);  $n_D^{20}$  1.4763. The iodomethylate of I melts at 153-154° C (ex methanol). Found, %: C 40.99; 40.91; H 6.65; 6.62; N 9.26; 9.53; I 43.26; 43.37%. Calculated for  $\text{C}_9\text{H}_{19}\text{N}_2$ , %: C 40.80; H 6.48; N 9.52; I 43.20%.

**N-(β-Cyanoethyl) isoindoline (II).** To 3.85 g (0.0322 mole) of isoindoline 2.56 g (0.048 mole) of acrylonitrile was added gradually and the heat in a sealed glass ampoule for 9 hr at 100° C. This yields 3 g (54.2%) of a white, crystalline product which darkens in the air, boils at 161-163° C (5 mm), and melts at 81°-83° C (ex acetone).

**N-Nitrosohexamethyleneimine (III).** To a mixture of 89.5 g (0.905 mole) of hexamethyleneimine, 75 ml of conc. HCl, and 26 ml of water were added at 70° C and under stirring during the course of 1 hr, 65 g of  $\text{NaNO}_2$  in 95 ml of water. Stirring was continued for 2 hr at 70° C, the upper layer was separated, and the aqueous layer was extracted with ether. The combined ether extracts were dried over calcined  $\text{Na}_2\text{SO}_4$  and distilled under vacuum. Yield 101.4 g (88.3%) III, boiling at 138° C (34 mm)\*, 119° C (17 mm);  $n_D^{20}$  1.4975. Found, %: N 22.04; 22.00%. Calculated for  $\text{C}_6\text{H}_{12}\text{N}_2\text{O}$ , %: N 21.75%.

**N-Nitrosoindoline (IV).** To a mixture of 8.25 g (0.0695 mole) of isoindoline, 8 ml of conc. HCl, and 2 ml of water at 80° C a solution of 10 g of  $\text{NaNO}_2$  in 15 ml of water was added with stirring in the course of 30 min. The precipitate was filtered off and recrystallized from alcohol. Yield 4.9 g (59.5%) of IV, mp. 95°-96° C\*\*.

**N-Aminohexamethyleneimine (VII).** To a solution of 26 g of III in 50 ml of dry ether, a suspension of 7.6 g of  $\text{LiAlH}_4$  in 200 ml of dry ether was added with stirring at a temperature about 38° C, the mixture boiled for 2 hr and allowed to stand overnight. After adding in succession 8 ml of water, 6 ml of a 20% solution of NaOH, and 28 ml of water, the mixture was filtered and the precipitate washed

\*According to the literature, III boils at 136-138° C (34 mm) [8].

\*\*According to the literature, IV melts at 95-97° C [9].

Characteristics of the Amines Obtained

Compound	Hydrogen conditions			Mp., ° C	Bp., ° C (pressure, mm)	$n_D^{20}$	Yield, %
	Temperature, ° C	Initial pressure, atm	Catalyst				
V	100	155	Co	—	116 (4)	1.4831	97.8
VI	80	100	Ni	98-99	—	—	74.5
VII	120	125	Ni	—	95 (55)	1.4853	61.2
VIII	95	100	Co	—	110 (26)	1.5670*	65.0

\*  $n_D^{19}$

with ether. By distillation were eliminated 6.7 g of unreacted III and 4.85 g (21%) of amine VII, boiling at 95° C (55 mm),  $n_D^{20}$  1.4853\*.

**Catalytic Reduction.** The reaction was conducted in a revolving steel autoclave with a volume of 0.25 liter. The catalyst was separated by decantation and the solution was distilled under vacuum. The purity of the amines was controlled by titrating a sample of the amines in aqueous alcohol with 0.1 N HCl. The table summarizes the hydrogenation conditions, the yields and the constants of the amines obtained.

## REFERENCES

1. J. H. Haury, *Arch. Pharm.*, **295**, 728, 1962.
2. A. L. A. Boura, F. C. Copp and A. F. Green, *Nature*, **195**, 1213, 1962.
3. E. Kh. Zolotarev, P. S. Bataev, V. I. Devyatova, *VSh, biol. nauki*, **4**, 16, 1961; *nauki*, **4**, 16, 1961.
4. E. Kh. Zolotarev, Yu. I. Kuznetsov, *Vestn. MGU, ser. VI, biol., pochvoved.*, **16**, 38, 1961.
5. E. Kh. Zolotarev, V. G. Mitrofanov, L. G. Yudin, N. B. Styashkina, *Vestn. MGU, ser. VI, biol., pochvoved.*, **16**, 4, 58, 1961.
6. A. N. Kost, P. B. Terent'ev, V. M. Byn'ko, *Vestn. MGU, ser. khim.*, **4**, 195, 1959.
7. *Amer. Pat.* 3041331, 1962, *C. A.*, **58**, 9037c, 1963.
8. A. P. Terent'ev, A. N. Kost, *ZhOKh*, **18**, 510, 1948.
9. F. Dürung, *Ber.*, **28**, 607, 1895.

---

\*According to the literature, VII boils at 94–96° C (55 mm) [8].

6 March 1966      Zelinskii Institute of Organic Chemistry, AS USSR, Moscow